

On the use of the ^{229m}Th isomers from the alpha decay of ^{233}U in a hydrochloric acid solution for a nuclear clock

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In isomeric transition the ^{229m}Th nuclei emit photons of 8.4 eV energy and a frequency of 2020.409 THz, which can serve as a reference frequency for the most accurate clocks. Isomeric transitions for ^{229m}Th atoms without valence electrons (ionization degree $n = 4+$) occur only by photon emission, at $n = 1+, 2+, 3+$ transitions mainly occur through an electron bridge, in neutral ^{229m}Th atoms, electron conversion mainly occurs. Initially, ^{229m}Th isomers were obtained in 2% of the alpha decay of ^{233}U nuclei. But in 2024, isomers were first obtained by resonant laser irradiation of ^{229}Th nuclei in the ground state in crystals where the Th atoms were in the form of ions. For the ^{229m}Th isomers obtained in this way, in [1] and in subsequent works, emission of isomeric transition photons was observed.

Laser excitation of isomers allows the development of nuclear clocks similar to atomic ones. The ^{229}Th nuclei in the ground state are embedded in a crystal, irradiated with a laser with a tunable frequency, and the resonance frequency is found based on the maximum excitation of the isomers, which is the reference frequency for the clock (see, e.g. [2]). However, unlike short-lived excited atomic states, the ^{229m}Th isomer lives in a crystal for more than 10 minutes, and when changing the laser frequency to search for resonance, it is necessary to wait for the decay of the previously excited ^{229m}Th isomer or replace the target.

In view of this, it is interesting to consider the possibility of amplifying the emission of photons from a source with the ^{229m}Th isomers and using this frequency-stable optical signal in a clock. For example, such a source could be ^{229m}Th isomers excited by a laser in a crystal. It is also promising to try a weaker, but also much simpler source with the ^{229m}Th isomers obtained by alpha decay of ^{233}U in a hydrochloric acid solution, where atoms with isomeric ^{229m}Th are in the form of $4+$ ions and the only channel for the isomers decay is photons emission [3]. In this way it is possible to observe the emission of photons by ^{229m}Th isomers, and not the luminescence of the solution under the action of radiation from other daughter products of U decay. Radiochemical separation of freshly formed Th from a solution with 100 mg of ^{233}U gives a source emitting ^{229m}Th photons with an intensity of ~ 1000 ph/s [3]. This intensity is sufficient for resonance amplification, see for example [4] about photon-level broadband spectroscopy with two frequency combs.

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